

Study of geopolymer paste Resulting From Alkali Activated Class F Black Fly ash With the different Combination Approach Of KOH And Na_2SiO_3

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ABSTRACT: This research paper involves the development of alkali activated fly ash paste resulting from different combination approach of KOH and Na_2SiO_3 as prime activator. In this experimental study an investigation was carried out to check the compressive strength, sorptivity water absorption & apparent porosity of the specimens. The type of fly ash, incorporated here for checking its physical and chemical properties of it, is class F (black). The combination of KOH and Na_2SiO_3 has been used as alkali activator. The basic objective of this paper is to study the development of various physical properties. The below investigation depends on same fly ash (class F) and difference on the K_2O content in activator fly ash.

KEYWORDS: Geopolymer, Activator, Fly Ash, Compressive Strength, Sorptivity

I. INTRODUCTION

Geopolymer is a kind of alkali aluminosilicate cementitious material, commonly known as fly ash concrete, having superior mechanical, chemical and thermal properties and with significantly lower CO_2 production [1]. Readily dissolved pozzolanic compound or source of silica and alumina in the alkaline solution is a renowned source of geopolymer [2]. Geopolymer has significant potential in a few aggressive situations where Portland cement concretes are vulnerable in nature [3]. Geopolymer materials tend to show high early strength, better durability and also depict almost no alkali-aggregate reaction [4]. These materials can be a far better replacement of cement in near future [5]. Low calcium fly ash based geopolymers have been reported to show outstanding compressive strength with performance, exposed to different acids and sulphate solution [6-12]. The earlier research on geopolymer includes that the process of geopolymer involves dissolution, orientation and poly condensation [13]. Various parameters like water content, alkali percentage, silicate modulus have a significant effect on geopolymer [14]. The earlier study suggests increment in compressive strength directly proportional to increase curing time and temperature [15]. In addition to that the composition and temperature on the properties of fly ash based geopolymer shows an outstanding effect on the compressive strength of geopolymer samples [16]. The below investigation is focussed on the activation of same kind of fly ash (class F) by different type of alkali combination.

II. EXPERIMENTAL

A. Materials Properties

Class F fly ash (black) is used here. The samples of fly ash were collected from Kolaghat Thermal Power Plant near Kolkata, India. Table-1 provides the chemical composition of the fly ash. The samples of fly ash has been sieved by 75 micron and specific gravity of above samples were checked which is 2.04. Laboratory grade sample of Potassium Hydroxide has 84% purity and supplied by Loba Chemie Ltd, India. Sodium silicate solution ($\text{SiO}_2 = 26.5\%$, $\text{Na}_2\text{O} = 8\%$ and 65.5% water) was supplied by Loba Chemie Ltd, India, which is carrying silicate modulus of 3.31. The alkali activator solution was prepared by mixing sodium silicate, potassium hydroxide and 32% water of fly ash. The solution (Alkali Hydroxide and water) was left for 24 hours. For class F (black) fly ash was $\text{SiO}_2/\text{X}_2\text{O}$ ratio (Here X_2O indicates the summation of K_2O and Na_2O in the activator solution) and $\text{SiO}_2/\text{K}_2\text{O}$ ratio was maintain as 0.77 and 1 respectively. Again it had X_2O content equal to 10.42% and 8.0% of fly ash for samples GPBF1 and GPBF2 respectively. Water to fly ash ratio was 0.32.

Table-1: Chemical analysis report of Fly ash

Chemical composition	Class F fly ash %
SiO ₂	51.3
Al ₂ O ₃	30.5
Fe ₂ O ₃	6.7
CaO	3.5
MgO	1
K ₂ O	0.86
Na ₂ O	1.2
SO ₃	3.1
Loss on ignition	0.6

B. Preparation of specimens and testing

Total mixing process was manually operated with predetermined quantity of activator solution for 5 minutes. Before mixing the sample fly ash was sieved by 75 microns. The activator is prepared by mixing KOH and Na₂SiO₃ together. Then the mixing was done manually. The mix was poured into a wooden mould, having dimension of 50mm each side. Table vibration was provided to eliminate any entrapped air. Then the cubes were cured in an oven at 85°C for a period of 48 hours [17]. After that, the testing specimens were removed from the moulds and stored at a dry place at room temperature. Mix data of the present study are given in the Table-2. After 3 days, the geopolymer specimens were tested for its compressive strength, apparent porosity and sorptivity tests. In support of each data point, twelve specimens were tested.

Table-2: Details of Geopolymer Specimens

Sample ID	K ₂ O content in Activator (%) of fly ash	In activator silicate modulus (SiO ₂ /K ₂ O)	In activator equivalent silicate modulus (SiO ₂ /X ₂ O)	Type of specimen	Water / fly ash ratio	Curing temp. and duration
GPB1	8	1	0.77	Paste	0.32	85°C and 48 hrs
GPB2	5.58	1.43	1	Paste	0.32	85°C and 48 hrs

Table-3: Combination of Different Alkali Oxide

Sample Id	*X ₂ O %	Na ₂ O %	K ₂ O %	Na ₂ O/ K ₂ O
GPB1	10.42	2.42	8	0.30
GPB2	8	2.42	5.58	0.43

*Here X refers to alkali Cation. For GPB1 and GPB2 samples X₂O indicates combination of Na₂O & K₂O. Below the details proportion has been held in Table-3.

III. RESULT AND DISCUSSION

A. Compressive strength

The random nature of samples related with breakage in few areas of geopolymer was chip of early to final failure. So here the samples with a distinct break were considered as the successful samples. The compressive strength of the geopolymer paste was determined after 48hrs of heat curing & 3 days of rest period respectively from manufacture. 12 samples of each series were tested for compressive strength in a compression testing machine and the average is reported. Compressive strength of geopolymer samples are presented in Fig-1. The result proved that the samples with higher percentage of K₂O shows better result among the two trials. The GPB1 sample indicates highest compressive strength of 36MPa after 3 days.

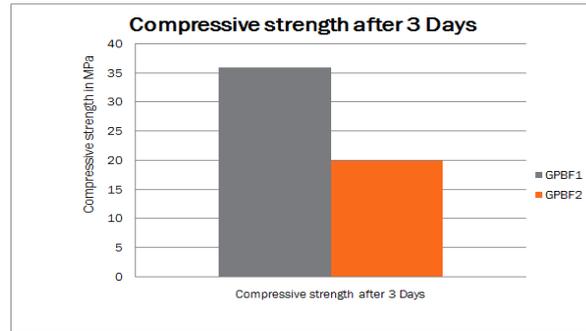


Fig-1. Compressive strength after 3 Days

B. Water sorptivity

Sorptivity is reflected to be a vigorous thing related with stability of ordinary cement concrete specimens [18]. Liquid tends to access into a non-saturated cement concrete because of sorption which is obviously compelled by the capillary forces. In this below study the curves for cumulative absorption of water scheme beside square root of time as shown in Fig-2. It is obvious from the Fig-2 that cumulative sorption is highest for GPB2 specimen. Similarly GPBF1 indicates least cumulative sorption of water. The maximum value of sorptivity indicates lowering in mean median size of pore which in fact indicates incomplete reaction.

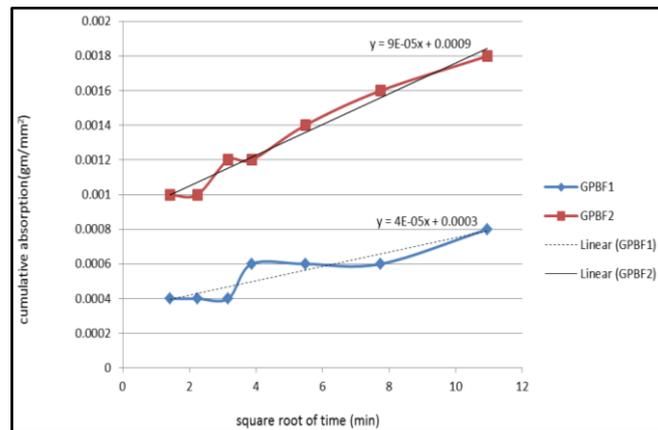


Fig-2. Water sorptivity of geopolymer samples

C. Apparent porosity and Water absorption

Apparent porosity and water absorption of the geopolymer specimens have been determined as per previous study [19]. Apparent porosity and water absorption of the specimens are represented in Table-4. The value of GPB1 and GPB2 specimen has the apparent porosity of 5.9% and 8.5% respectively. GPB2 Specimens have growing water absorption than GPB1 sample. These phenomena possibly leave a positive impact on the pore morphology of geopolymer in this investigation. Similarly in the same trend the value of apparent porosity is much lower for GPB1 sample. So it can be concluded that geopolymer specimens with high percentage of potassium hydroxide content tends to show dense characteristics for geopolymer specimens.

Table-4: Results of water absorption, apparent porosity

SAMPLE ID	Dry wt. of sample	Wt. of sample at wet condition	Wt. of sample at suspended condition	Apparent porosity	Water absorption
GPB1	193	200	82	5.93	3.62
GPB2	176	185	80	8.57	5.11

Sample ID	Water absorption (%)	Apparent porosity (%)
GPB1	3.6	5.9
GPB2	5.1	8.5

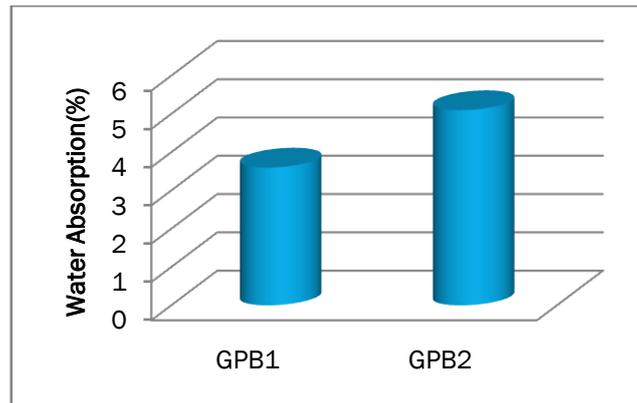


Fig-3: graphical presentation of water absorption

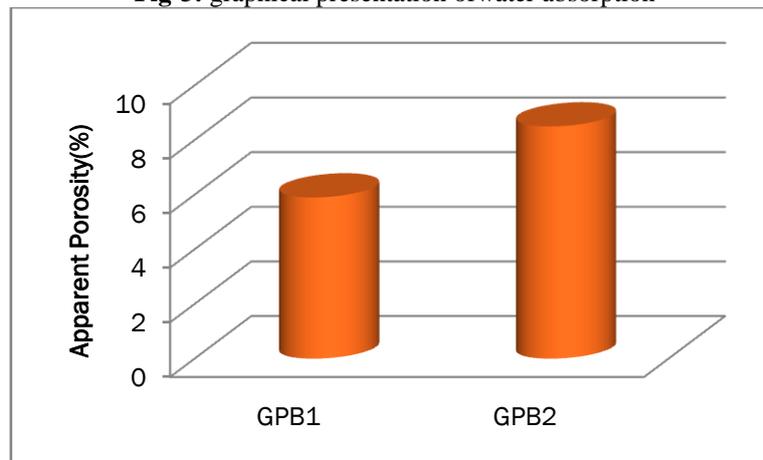


Fig-4: graphical presentation of apparent porosity

IV. CONCLUSION

Based on the results of the present experimental investigation, following conclusions are drawn.

- ❖ The compressive strength is higher for specimens GPBF1. It implies higher activation in the presence of higher content of silicon in fly ash in the midst of high alkalinity.
- ❖ The water sorptivity is lowest for GPBF1 samples. This is due to less capillary rise which is impossible unless the pores are discontinuous.
- ❖ The water absorption is lowest for GPBF1 samples. This can be spotted over strength and durability character.
- ❖ The apparent porosity is lowest for GPBF1 samples. This characteristic supports the presence of impermeable pores within the geopolymeric structure.

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